

Validated HPLC Method for the Determination of Fenofibric Acid in Rat Plasma and its Application to a Comparative Pharmacokinetic Study of Prodrugs JWU102 and Fenofibrate

Research Article

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Abstract

In this study, a sensitive and reliable method for the quantitation of fenofibric acid in rat plasma was developed and validated using high performance liquid chromatography (HPLC). The plasma samples were prepared by deproteinization and sildenafil was used as an internal standard. Chromatographic separation was achieved using a reversed-phase (C18) column. The mobile phase, 0.02 M ammonium acetate buffer: acetonitrile (35:65, v/v), was run at a flow rate of 1.0 mL/min, and the column eluent was monitored using an ultraviolet detector at 280 nm at room temperature. The retention times of sildenafil (an internal standard), and fenofibric acid were approximately 5.9 and 7.7 min, respectively. The quantitation limit of fenofibric acid in rat plasma was 0.03 µg/mL. Pharmacokinetic parameters of fenofibric acid was evaluated after oral (at doses of 20 mg/kg) administration of JWU102 and fenofibrate in rats. The relative bioavailability of JWU102 was approximately 272.8% compared to fenofibrate.

Keywords: JWU102; Fenofibric Acid; HPLC; Bioavailability; Rats.

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Introduction

Cardiovascular disease (CVD) and coronary heart disease (CHD) are leading causes of death in worldwide. Risk factors for CHD include elevated low-density lipoprotein cholesterol (LDL-C), elevated triglycerides, and decreased concentrations of high-density lipoprotein cholesterol (HDL-C) [1]. Due to the overwhelming evidence generated from outcome trials, statins remain the primary method for lowering LDL-C and reducing the incidence of cardiovascular events in these patients [2]. Although statin therapy plays an important role in improving the lipid profile of patients, approximately 10% to 22% of individuals in clinical studies experience muscle pain during statin therapy [3]. Mixed dyslipidemia

characterized by low levels of HDL-C and high levels of triglycerides and LDL-C is highly prevalent in the general population, particularly in obese patients with metabolic syndrome [4]. If a patient at high risk has high triglycerides or low HDL-C, consideration can be given to combining a fibrate or nicotinic acid with an LDL-C lowering drug.

Fenofibrate acts by stimulating the activity of peroxisome proliferator-activated receptor-α (PPAR-α), a member of the PPAR subfamily of nuclear receptors that modulate the transcription of genes that regulate fatty acid and cholesterol metabolism [5]. Fenofibrate (Figure 1B), a prodrug, is pharmacologically inactive and undergoes rapid hydrolysis at the ester bond to form the active metabolite fenofibric acid (Figure 1C) [6]. However, fenofibrate is a neutral, lipophilic compound that is practically insoluble in water, making it challenging to consistently achieve therapeutic levels [7]. Thus, several different formulations of fenofibrate have been developed in an attempt to increase its overall solubility since its introduction in the United States.

The purpose of this study was to develop and validate a sensitive and reliable high-performance liquid chromatographic (HPLC) method of fenofibric acid. Based on the developed method, we determined the pharmacokinetic properties of fenofibric acid after oral administration of JWU102 and fenofibrate at a dose of 20 mg/kg to rats to evaluate the relative bioavailability of JWU102 in rats.

Materials and Methods

Materials

JWU102, fenofibrate, and fenofibric acid were supplied from Jungwon university (Chungbuk, South Korea). Sildenafil, an internal standard (IS) for the high-performance liquid chromatographic

Figure 1. Chemical structures of JWU102, fenofibrate, and fenofibric acid.

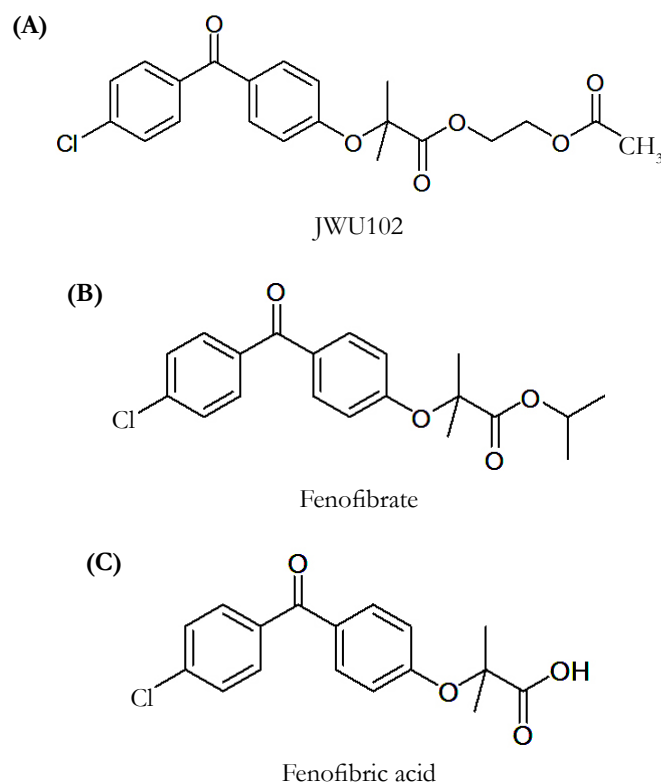
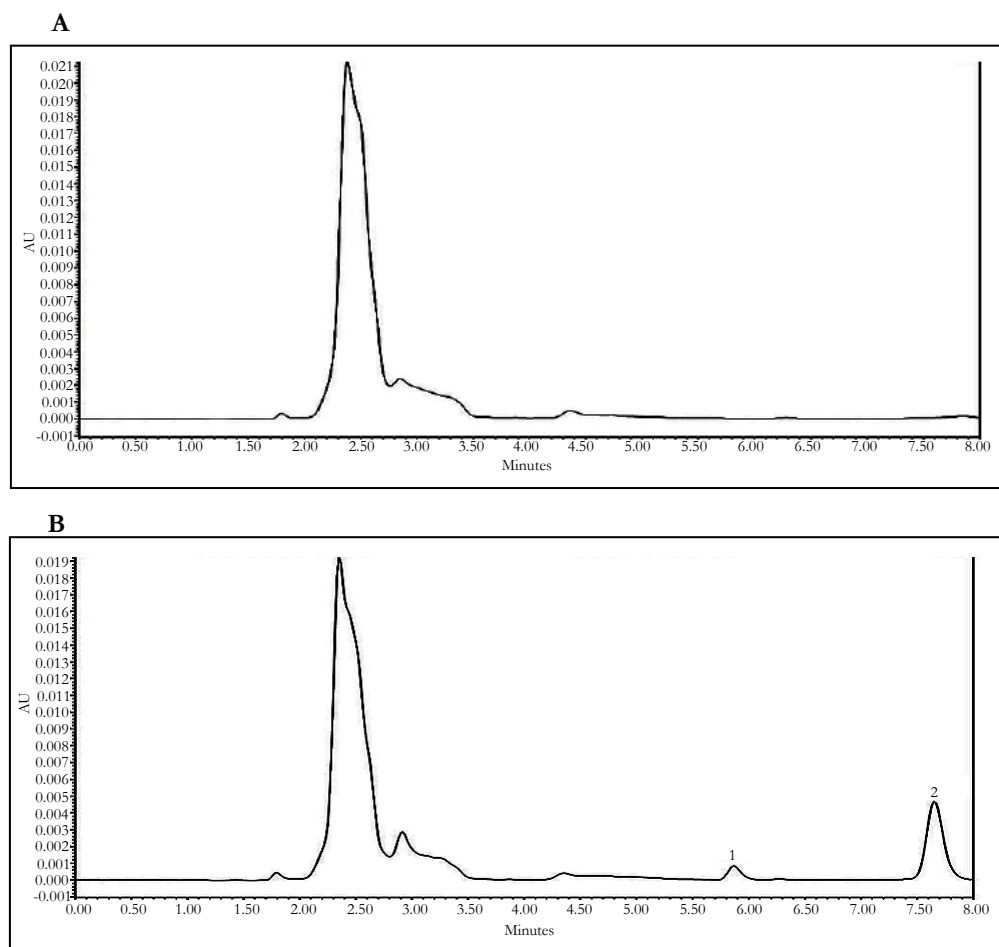
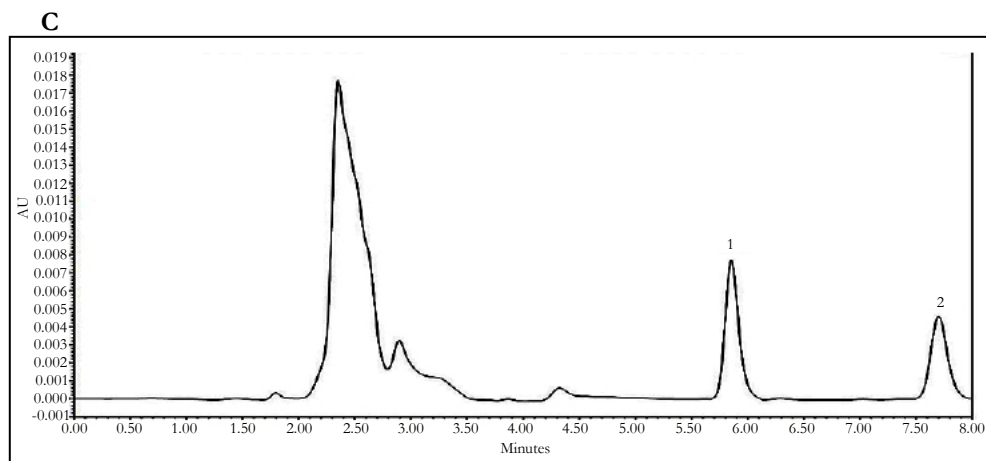


Figure 2. Representative HPLC chromatograms after deproteinization of drug-free rat plasma (A), drug-free rat plasma spiked with 0.03 µg/mL (LLOQ) of fenofibric acid and 6 µg/mL of IS (B), plasma collected 1 hr after oral administration of 20 mg/kg of fenofibrate to a male Sprague–Dawley rat (C). Peaks: (1): fenofibric acid (5.9 min); (2): IS (7.7 min).





(HPLC) analysis was purchased from Sigma–Aldrich Corporation (St. Louis, MO). Acetonitrile and methanol were products from Burdick & Jackson (Muskegon, MI, USA). Polyethylene glycol 400 (PEG 400) was a product from Showa Chemical Company (Tokyo, Japan). Other chemicals were of reagent grade or HPLC grade.

Animal Experiments

Male Sprague–Dawley rats, 6–8 week old and weighing 220–300 g, were purchased from the Samtako Bio Korea (Osan, South Korea). Rats were maintained in a Clean room at a temperature of between $23\pm 2^\circ\text{C}$ with 12-h light (07:00–19:00) and dark (19:00–07:00) cycles, and a relative humidity of $55\%\pm 5\%$. Rats were housed in metabolic cages (Tecniplast, Varese, Italy) under filtered pathogen-free air and with food (Sam Yang Company, Pyeongtaek, South Korea) and water available ad libitum. The rats were fasted overnight before drug administration and for 4 hr after dosing. JWU102 was dissolved in PEG400: distilled water=1:1 (v/v) to make a concentration of 5 mg/mL. The rats were placed in a restrainer and were orally administered a dose of 20 mg/kg. Blood was collected in a heparinized tube at the pre-dose stage, and at 0.25, 0.5, 1, 2, 4, 6, 8, 10, and 12 h after oral administration. Plasma was harvested after centrifugation at 3,000 rpm and 4°C for 10 min and stored frozen at -70°C until it was analyzed.

Preparation of calibration standards and quality control samples

Stock solutions of fenofibric acid (1 mg/mL) were prepared in methanol. Appropriate dilutions of the stock solutions of fenofibric acid were made with methanol (0.003, 0.005, 0.01, 0.02, 0.05, 0.1, 0.2, 0.5, or 1 mg/mL). Standard solutions of fenofibric acid in rat plasma were prepared by spiking with an appropriate volume (10 μL /mL of plasma) of the diluted stock solutions, giving final concentrations of 0.03, 0.05, 0.1, 0.2, 0.5, 1, 2, 5, or 10 μg /mL for plasma. The IS working solution was prepared by dissolving sildenafil in acetonitrile to give a final concentration of 10 μg /mL.

Preparation of plasma samples

A 50 μL aliquot of sample was deproteinized with a 75 μL of acetonitrile containing 1 μg /mL sildenafil (an IS). After vortex-mixing and centrifugation at 3,000 rpm for 10 min, the supernatant was transferred into a vial and a 20 μL aliquot was injected directly onto the HPLC column.

HPLC analysis

The HPLC system consisted of a Gilson-234 autosampler (Gilson, Middleton, WI, USA), a Gilson 307 pump (Gilson), a Capcell PACK (C_{18}) column (250 mm \times 4.6 mm, i.d.; particle size, 5 μm ; Shiseido, Tokyo, Japan), a model UV-118 UV/VIS detector (Gilson), and a model Gilson unipoint system software (Gilson). The mobile phase, 0.02 M ammonium acetate buffer:acetonitrile (35:65, v/v), was run at a flow rate of 1.0 mL/min, and the column eluent was monitored using an ultraviolet detector at 280 nm at room temperature. The retention times of IS and fenofibric acid were approximately 5.9 and 7.7 min, respectively.

Analytical method validation

The analytical method was validated with regards to its specificity, linearity, intra- and interday precision and accuracy, matrix effect, and stability according to the US Food and Drug Administration's "Guidance for Industry, Bioanalytical Method Validation, 2001 [8]"

Pharmacokinetic and statistical analysis

The total area under the plasma concentration-time curve to the last time (AUC_{last}), the maximum plasma concentration (C_{max}), the time to reach C_{max} (T_{max}), and the half-life ($T_{1/2}$) were estimated using noncompartmental calculations carried out within WinNonlin™ 5.2 (Pharsight, Sunnyvale, CA, USA). All data are expressed as the mean \pm standard deviation (SD). The statistical significance of the differences between the 2 groups was analyzed using Student's *t*-tests carried out within SPSS (IBM, Yorktown Heights, NY, USA). A *p* value of <0.05 was considered statistically significant.

Results and Discussion

Development and validation of the HPLC method

Representative chromatograms of the deproteinized drug-free rat plasma, drug standards in rat plasma spiked with 0.03 μg /mL (LLOQ) of fenofibric acid and 6 μg /mL of IS, and plasma collected 1 hr after oral administration of 20 mg/kg of fenofibrate to a male Sprague–Dawley rat were shown in Figure. 2 (A), (B), and (C), respectively. No interferences from endogenous substances were observed in the blank rat plasma samples. The

retention times of IS and fenofibric acid were 5.9 and 7.7 min, respectively. The analytical method used was linear over the range of 0.03–10 µg/mL, with correlation coefficients (r values) greater than 0.9997. The lower limit of quantitation was 0.03 µg/mL with relative standard deviation (RSD) values less than 20% and relative errors within ± 20%. Intra- and inter-day accuracies (as relative error values) ranged between 1.0% and 11.5% and intra- and inter-day precision (as RSDs) were 3.0–10.1% for all QC (quality control) samples, with the result that they all met the criteria for bioanalysis method validation (Table 1).

The matrix effect, recovery, and process efficiency values for fenofibric acid and IS in rat plasma are provided in Table 2. The recovery was, on average, more than 90% for both compounds. Fenofibric acid was found to be stable under various conditions, whether in the plasma or in the stock solution, and the detailed stability data are presented in Table 3. In summary, the HPLC method developed in the current study was found to be suitable for the quantitation of fenofibric acid in rat plasma with acceptable specificity, linearity, accuracy, precision, and stability. On the basis of this HPLC method, fenofibric acid concentrations in rat plasma were determined and pharmacokinetic parameters were calculated.

Comparative pharmacokinetics of JWU102 and fenofibrate after oral administration to rats

Plasma samples were collected after the oral administration of JWU102 and fenofibrate and the concentrations of fenofibric acid were determined using the validated HPLC method. Figure.

3 shows the mean plasma concentration-time curves for fenofibric acid after the oral administration of JWU102 and fenofibrate in rats; the pharmacokinetic parameters are presented in Table 4. After oral administration of JWU102, absorption of fenofibric acid from the rat portal vein was slow and the plasma concentrations of fenofibric acid declined in delay-released manners with a T_{max} and a half-life of 6.0 hr and 7.9 hr, respectively. The estimated total body clearance was 33.0 mL/min/kg which was considerably smaller than the reported cardiac output of 295 mL/min/kg based on blood data [9] in rats. This suggests that the first-pass effect of JWU102 in the lung and heart could be almost negligible in rats.

The maximum plasma concentrations of fenofibric acid were achieved 6.0 and 4.0 hr after oral administration for JWU102 and fenofibrate, respectively. The C_{max} values were 39.19 ± 6.65 and 22.78 ± 3.68 µg/mL, and the AUC_{last} values were 343.13 ± 61.79 and 125.80 ± 22.42 µg•h/mL for JWU102 and fenofibrate, respectively. The C_{max} value of JWU102 was 1.7 times higher than that of fenofibrate and AUC_{last} value of JWU102 was 2.7 times higher than fenofibrate. JWU102 appeared to have been significantly well absorbed more than fenofibrate from the gastrointestinal tract. The p-value of C_{max} and AUC_{last} between 2 groups were 0.020 and 0.005, respectively. This suggests that JWU102 was well absorbed via peptide transporter 1 from the gastrointestinal tract [10, 11]. The considerable effect of JWU102 on peptide transporter 1 is being tested based on the homogenate studies. The relative bioavailability of JWU102 was 272.8% compared to fenofibrate. This implies that the dosage of JWU102 could be smaller than fenofibrate about 2.5–3.0 folds in clinical studies.

Table 1. Intra- and inter-day precision and accuracy for fenofibric acid in rat plasma QC samples.

Nominal conc. (ng/mL)	Measured conc. (ng/mL)	Coefficient of variation (%)	Relative error (%)
Intra-day (n=6)			
30	29.5 ± 2.4	8.1	-1.7
100	106.3 ± 5.5	5.2	6.3
1000	1109.2 ± 33.7	3	10.9
7500	7827.4 ± 314.9	4	4.4
Inter-day (n=18, 6 runs per day)			
30	31.8 ± 3.2	10.1	6.0
100	111.5 ± 8.3	7.4	11.5
1000	1080.6 ± 55.2	5.1	8.1
7500	7427.9 ± 438.3	5.9	-1.0

Data represent mean ± SD.

Coefficient of variation (%) = (SD/mean) × 100

Relative error (%) = ((Measured conc. - Nominal conc.) / Nominal conc.) × 100

Table 2. Matrix effect, recovery, and process efficiency data for fenofibric acid and sildenafil in rat plasma.

Concentration (ng/mL)	Matrix effect (%) (B/A×100)	Recovery (%) (C/B×100)	Process efficiency (%) (C/A×100)
Fenofibric acid			
100	72.5 ± 12.9	96.0 ± 2.9	69.3 ± 11.2
1000	71.6 ± 4.3	93.9 ± 6.3	67.3 ± 6.0
7500	82.6 ± 3.8	86.5 ± 2.1	71.4 ± 3.5
Sildenafil			
10000	73.8 ± 1.9	91.6 ± 2.6	67.6 ± 2.1

A, Peak area of analytes in mobile phase

B, Peak area of analytes spiked after extraction

C, Peak area of analytes spiked before extraction

Table 3. Stability of fenofibric acid in rat plasma and stock solutions (n=3).

Nominal conc. (ng/mL)	Duration	Measured conc. (ng/mL)	Relative error (%)
Short-term stability (at room temperature, RT)			
100	4 h	102.7 ± 6.5	2.7
1000		1030.5 ± 34.8	3.1
7500		7625.5 ± 323.8	1.7
Long-term stability (at -80°C)			
100	7 days	109.8 ± 10.1	9.8
1000		1076.2 ± 45.6	7.6
7500		7703.2 ± 418.3	2.7
Freeze and thaw stability			
100	3 cycles	105.4 ± 8.2	5.4
1000		1070.2 ± 40.7	7
7500		8010.3 ± 380.7	6.8
Auto-sampler stability (at 4°C)			
100	24 h	103.6 ± 8	3.6
1000		1053.7 ± 51.2	5.4
7500		7865.5 ± 638.8	4.9
Stock solution			
500	2 h at RT	510.23 ± 5.3	2
	11 days at 4°C	489.7 ± 3.6	-2.1

Data represent mean ± SD

$$\text{Relative error (\%)} = ((\text{Measured conc.} - \text{Nominal conc.}) / \text{Nominal conc.}) \times 100$$

Figure 3. Mean plasma concentration–time profiles of fenofibric acid after oral administration of JWU102 (Δ) and fenofibrate (▲) at a dose of 20 mg/kg to rats. Each point represents the mean ± standard deviation (n = 5).

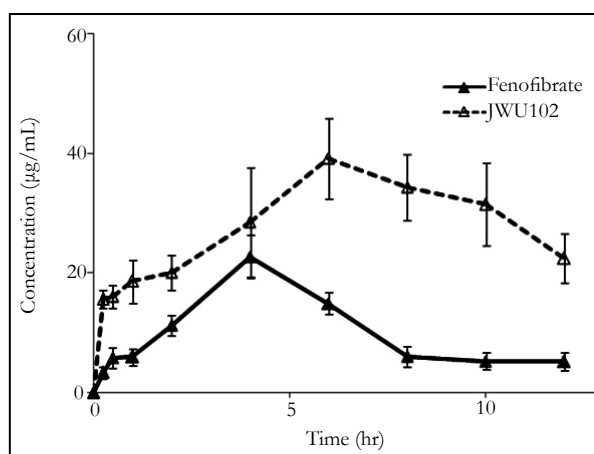


Table 4. Pharmacokinetic parameters of fenofibric acid after a single oral administration of JWU102 and fenofibrate at a dose of 20 mg/kg to male rats.

Parameters	JWU102 (n=5)	Fenofibrate (n=5)
AUC ₀₋₁₂ (µg·hr/mL)	343.13 ± 61.79	125.80 ± 22.42
C _{max} (µg/mL)	39.19 ± 6.65	22.78 ± 3.68
T _{max} (hr)	6.00 ± 0.00	4.00 ± 0.00
T _{1/2} (hr)	7.89 ± 1.49	3.46 ± 0.78

Data represent mean ± SD (n=5).

AUC: Area under the curve to the collected time point (µg·hr/mL).

C_{max}: Peak plasma concentration (µg/mL)

T_{max}: Time to reach peak plasma concentration (hr)

T_{1/2}: Elimination half life (hr)

Conclusion

The developed HPLC method was suitable for the quantification of fenofibric acid in rat plasma with acceptable specificity and linearity. Intra- and inter-day accuracies and intra- and inter-day precision met the criteria for bioanalysis method validation. Fenofibric acid was found to be stable under various conditions, whether in the plasma or in the stock solutions. The recovery of fenofibric acid from the processed plasma indicated that there is no significant matrix effects. According to this method, the concentrations of fenofibric acid in rat plasma were determined and pharmacokinetic parameters were calculated. The pharmacokinetic parameters of JWU102 was higher than that of fenofibrate and relative bioavailability was 272.8% compared to fenofibrate. This implies that the dosage of KB002 could be smaller than fenofibrate about 2.5–3.0 folds in clinical studies.

Acknowledgement

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